



9-1-2011

Lotus effect toy

Said Shakerin

University of the Pacific, sshakerin@pacific.edu

Follow this and additional works at: <https://scholarlycommons.pacific.edu/soecs-facarticles>

 Part of the [Mechanical Engineering Commons](#)

Recommended Citation

Shakerin, S. (2011). Lotus effect toy. *Physics Teacher*, 49(6), 346–347. DOI: [10.1119/1.3628257](https://doi.org/10.1119/1.3628257)
<https://scholarlycommons.pacific.edu/soecs-facarticles/85>

This Article is brought to you for free and open access by the All Faculty Scholarship at Scholarly Commons. It has been accepted for inclusion in All Faculty Articles - School of Engineering and Computer Science by an authorized administrator of Scholarly Commons. For more information, please contact mgibney@pacific.edu.

Lotus Effect Toy

Cite as: Phys. Teach. **49**, 346 (2011); <https://doi.org/10.1119/1.3628257>

Published Online: 24 August 2011

Said Shakerin



View Online



Export Citation

ARTICLES YOU MAY BE INTERESTED IN

[Serious Fun: Using Toys to Demonstrate Fluid Mechanics Principles](#)

The Physics Teacher **52**, 332 (2014); <https://doi.org/10.1119/1.4893085>

[Cartesian Diver Plus](#)

The Physics Teacher **58**, 84 (2020); <https://doi.org/10.1119/1.5144783>

[Is the lotus leaf superhydrophobic?](#)

Applied Physics Letters **86**, 144101 (2005); <https://doi.org/10.1063/1.1895487>



COMMUNITIES

Network with Peers, Resource Library, Message Boards, Meetings, and more!

Lotus Effect Toy

Said Shakerin, University of the Pacific, Stockton, CA

This is a short article about Aqua Drop, a toy that employs a superhydrophobic surface. After defining superhydrophobic surface, its occurrence in nature, and its importance in science and technology, I describe the toy, followed by several student activities that can be performed with the toy and easily found natural objects.

Background

Several plants and insects exhibit remarkable surfaces (e.g., leaves and wings) that combine chemical and physical attributes that cause water falling on them to effectively bead up and roll off. This ability serves for cleaning and survival purposes. These surfaces are waxy, making them hydrophobic (“water hater”). They also have nanometer-scale roughness elements that make them superhydrophobic. The difference between hydrophobic and superhydrophobic is characterized by the contact angle shown in Fig. 1.¹ If this angle is between 90° to 150° , the surface is called hydrophobic; for angles higher than 150° , the surface is classified as superhydrophobic. (Between about 5° to 90° , the surface is hydrophilic [“water lover”] and below 5° is considered superhydrophilic.)

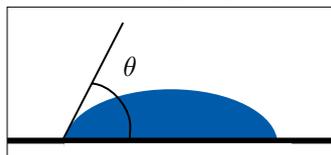


Fig. 1. Water drop on a surface (θ is the contact angle).

Wilhelm Barthlott, a professor of botany in Germany, discovered the mechanism of superhydrophobicity by observing the leaves of a lotus plant under a scanning electron microscope.² The term “lotus effect” was coined to exemplify superhydrophobicity of the plant. That discovery opened up new research and development in search of self-cleaning surfaces, low-drag coatings, and innovative fluid handling devices.³ Currently, outdoor paints are marketed that make the façade of a building self-cleaning. That is, during a rain shower

the accumulated dust particles on the surface are washed off by being collected on the rain drops as they bead up and roll off the surface, similar to what happens in nature on the lotus leaf. Furthermore, there are now fabrics available that allow ketchup, coffee, or wine spills to easily roll off of them. Several research teams around the world are investigating the creation of switchable surfaces to make them superhydrophobic or superhydrophilic depending on the application. This is of interest to designers of micro-sized fluidic devices.^{4,5} A recent review of superhydrophobic and superhydrophilic plant surfaces as applied to biomimetic materials is available.⁶

Toy and student activities

Aqua Drop is a relatively new toy that vividly exhibits the lotus effect.⁷ Shown in Fig. 2, the toy is made of a shallow plastic box containing a maze whose surface is superhydrophobic. Although the surface looks smooth, it has roughness elements on the order of nanometers, and a special polymer gives it a waxy shine. The maze includes several hemispherical depressions. An eye dropper is supplied with the toy and a small hole on the top of the box allows insertion of drops onto the maze with the eye dropper. An outlet hole is at the lower side of the box to drain water after using the toy. The objective is to bring to and park a water drop in each of the depressions. This is not easy! Even a tilt of a few degrees causes the water drops to roll on the surface. A similar toy that used an enclosed drop of mercury was available in the 1990s (see Fig. 3).

Besides playing with the maze to place a drop in each of the depressions, one can create a very large drop of water on the surface and then observe its movement around the maze by slightly tilting the toy back and forth and sideways. If the toy is shaken, the large drop breaks down into small drops, which then can be recombined by tilting the toy toward a corner and observing the coalescence of smaller drops into a large one. Another neat activity is to release two drops, one dyed with food coloring, into the toy and then observe their coalescence and the subsequent color dilution. Also, students can gather

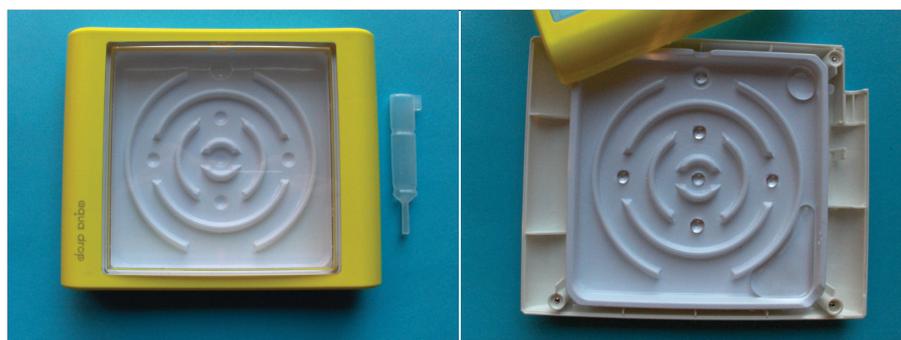


Fig. 2. Left, Aqua Drop with the included eye dropper. Right, small water droplets are seen in depressions and large drops in the right corners. (Cover retracted for clarity.)



Fig. 3. Maze toy with the mercury drop at the bottom. (Photo courtesy of Christian Ucke.)

natural materials such as bird feathers, plant leaves, flower petals, and fuzzy fruits such as peaches, and observe the differences in contact angles by placing a water drop on each.

The toy and activities described above allow students to gain an appreciation for superhydrophobic phenomenon, a topic of current interest in scientific and engineering fields.

References

1. L. Courbin and H. A. Stone, "Your wetting day," *Phys. Today* **60** (2), 84–85 (2007).
2. C. Neinhuis and W. Barthlott, "Characterization and distribution of water repellent, self-cleaning plant surfaces," *Ann. Bot.* **79**, 667–677 (1997).
3. M. Ma and R. Hill, "Superhydrophobic surfaces," *Curr. Opin. Colloid Interface. Sci.* **11**, 193–202 (2006).
4. T. Krupenkin et al., "Reversible wetting-dewetting transitions on electrically tunable superhydrophobic nanostructured surfaces," *Langmuir* **23**, 9128–9133 (2007).
5. C. Dorrer and J. Ruhe, "Wetting of silicon nanoglass: From superhydrophilic to superhydrophobic surfaces," *Adv. Mater.* **20**, 159–163 (2008).
6. K. Kock and W. Barthlott, "Superhydrophobic and superhydrophilic plant surfaces: An inspiration for biomimetic materials," *Phil. Trans. R. Soc. A* **367**, 1487–1509 (2009).
7. Aqua Drop, available in four maze patterns, can be found online.

Said Shakerin is a professor of mechanical engineering at the University of the Pacific in Stockton, CA. He was educated at Arya-Mehr (now Sharif) University of Technology in Iran, Portland State University, Oregon State University, and Colorado State University. He is a registered mechanical engineer in California and served as department chairman in the 1990s but stepped down due to a medical condition. Among his interests are development of teaching tools to enhance students' learning and design of water fountains with special effects and other objects combining fluid dynamics and visual arts.

Department of Mechanical Engineering, University of the Pacific, Stockton, CA 95211; sshakerin@pacific.edu